Atmosphere

Earth's atmosphere

- Earth's atmosphere is a layer of gases surrounding the planet.
- The Earth is surrounded by a blanket of air, which we call the atmosphere. It reaches over 560 kilometers from the surface of the Earth.

Atmosphere:

- > Absorbs the energy from the Sun,
- Recycles water and other chemicals,
- > protects us from high-energy radiation and the frigid vacuum of space.
- > The atmosphere protects and supports life.

Earth's atmosphere

- Earth's atmosphere is made of a mixture of gases called air.
 - Nitrogen gas makes up about 78% of Earth's atmosphere.
 - The second most abundant gas is oxygen, which makes up 21% of Earth's atmosphere.
 - ➤ The third Argon (Ar, 0.9%).



> Carbon Dioxide (CO_2 , 0.03%).

Composition of the Atmosphere

The atmosphere is comprised of a variety of gases:

> Major Constituents (99%):

≻ Nitrogen (N): 78%

> Oxygen (O_2): 21%

> Trace Constituents:

- > Argon (Ar), about 0.9%
- > Water vapor (H_2O), up to 10000 ppmv
- \succ Carbon dioxide (CO₂), 350 ppmv
- > Ozone (O_3), near zero at the surface, up to 10 ppmv in the stratosphere
- > Methane (CH_4), 1.7 ppmv
- ➤ and others.....

ppmv = "parts per million by volume"

Nitrogen Cycle

- Nitrogen is important to protein which is found in the body tissues of all living things.
- Nitrogen is cycled through the soil and into plants and finally when living things die and decay.



Pressure in the atmosphere

- Atmospheric pressure is the force per unit area exerted into a surface by the weight of air above that surface in the atmosphere of Earth.
- The gas molecules closest to Earth's surface are packed together very closely.
- This means pressure is lower the higher up you go into the atmosphere.



Pressure in the atmosphere

- At sea level, the weight of the column of air above a person is about 9,800 Newtons (2,200 pounds)!
- This is equal to the weight of a small car.
- 1 atm = 14.69 pound/sq inch
 = 6.6 kg/ sq inch
 Other units of pressure are mm of Hg, millibar, pascal



Measuring Pressure

- A barometer is an instrument that measures atmospheric pressure.
- Long ago, mercury barometers were used
- Since mercury is a poisonous liquid, aneroid barometers are used today.





Pressure changes with altitude



ozone 🚱

stratosphere

thermosphere

mesosphere

troposphere

R Behera

The atmosphere has four layers ≻Thermosphere

≻Mesosphere

> Stratosphere

≻Troposphere



Troposphere

- Lowest and thinnest layer
 16 km at equator, 8 km at poles
- > 90% of the atmosphere's mass
- > Temperature decreases with altitude
 - ➢ 6°C per kilometer
 - ➢ Top of troposphere averages −50°C



Boundary between the troposphere, and the stratosphere is called the tropopause



View of troposphere layer from an airplane's window.

Stratosphere

- > Extends from 10 km to 50 km above the ground
- Less dense (less water vapor)
- > Temperature **increases** with altitude
- Almost no weather occurrence
- Contains high level of ozone

> Ozone layer

> Upper boundary is called stratopause.



Mesosphere

> Extends to almost 80 km high

➤ Gases are less dense.



Temperature decreases as altitude increases.
 Gases in this layer absorb very little UV radiation.

Thermosphere

Above the mesosphere and extends to almost 600 km high

- Temperature increases with altitude
- Readily absorbs solar radiation





Reflects radio waves



- The four layers of the atmosphere include:
- 1. *the troposphere*, *where we live;*
- 2. the stratosphere, which contains the ozone layer;
- 3. the mesosphere, where meteors burn; and
- 4. the thermosphere, where satellites orbit Earth.



The exosphere begins at about 500 kilometers above Earth and does not have a specific outer limit.

Satellites orbit Earth in the exosphere.



The exosphere and ionosphere

- Communication on Earth depends on satellites.
- Satellites transmit information used for television shows, radio broadcasts, data and photos used in weather reports, and long distance telephone calls.



The ozone layer

In the 1970s, scientists noticed that the ozone layer in the stratosphere above Antarctica was thinning. The largest hole in the ozone layer ever observed. (September 24, 2006)



Chlorofluorocarbons & the ozone layer

- A group of chemicals called chlorofluorocarbons (or CFCs) were once commonly used in air conditioners, in aerosol spray cans, and for cleaning machine parts.
- In the London Agreement of 1991, more than 90 countries banned the production and use of CFCs except for limited medical uses.



Chlorofluorocarbons & the ozone layer

- The ozone layer absorbs the Sun's high-energy ultraviolet (UV) radiation and protects the Earth.
- In the stratosphere, the CFCs break down and release chlorine.
- The chlorine reacts with ozone molecules, which normally block incoming ultraviolet radiation.



Chlorofluorocarbons (CFCs) and Ozone Depletion



Particulates

NO

Emissions

so,

Acid rain occurs when oxides of sulfur and oxides of nitrogen are emitted into the atmosphere, undergo chemical transformations and are absorbed by water droplets in clouds.

Deposition

NH,

NO.

Hg

Effects

AI NH, Ca

NO, H SO,

R Behera

Effects of Acid Rain

- ➤Acidification of bodies of water
- Damage of vegetation
- ≻Damage to building materials, statues, etc.





GREENHOUSE EFFECT

The trapping of heat by gases in the atmosphere.

> Naturally occurring greenhouse gases:

- ➤ Water vapor
- Carbon dioxide
- ➤ Methane
- > Nitrous oxide
- ➢ Ozone

> Greenhouse gases that are not naturally occurring

- > *Hydro fluorocarbons* (HFCs)
- > Per fluorocarbons (PFCs)
- > Sulfur hexafluoride (SF₆)

Generated in a variety of industrial processes.

The Greenhouse Effect on Earth

Earth's atmosphere is slightly warmer than what it should be due to direct solar heating because of a *mild case of greenhouse effect...*

- The ground is heated by visible and (some) infrared light from the Sun.
- The heated surface emits infrared light.
- The majority of Earth's atmosphere (N₂ and O₂) are not good greenhouse gas.
- The small amount of greenhouse gases (H₂O, CO₂) traps (absorb and re-emit) the infrared radiation, increasing the temperature of the atmosphere...



Surface absorbs visible light and emits thermal radiation in infrared.

Some Solar radiation redirected by both the Earth & atmosphere

Solar radiation Passes through the Earth's atmosphere Some of the infrared radiation passes through the atmosphere. some is observed and reemitted in all directions by greenhouse gas molecules. This causes the earth surface and lower atmosphere to

warm

The majority of the radiation is absorbed by the Earth's Surface with it warms

SUN

Infrared radiation is realized from the Earth Surface EARTH

Atmosphere



Greenhouse Gases



Recent 124°F Highs (51°C) INDIA Worst hot spell 122°F > in 50 years (50°C) kills about 3,000 people BAHRAIN in a place used to 117°F a merciless sun, (47°C) 56 workers suffer heat exhaustion TEXAS A month of 100"-113°F ► plus temperatures kills more than (45°C) 120 people and CALIFORNIA destroys about Early August a third of the heat wave cotton crop brings blackouts and "spare the 110°F air" days to the (43°C) San Francisco Bay Area CYPRUS Worst heat wave in 40 106°F > years kills at (41°C) least 55 GERMANY people and sends 3,200 Record hot to hospitals streak produces severe smog, In some areas. 100°F vehicles without (38°C) antipollution devices are FRANCE banned Sun withers 20% of the grapevines in 95°F ► some areas of (35°C) Bordeaux SWITZERLAND Scorching heat sends kids home TIME Graphics by Ed Gabel from school

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PLANET WATCH

Global Warming: It's Here ...

Scientists are increasingly convinced that the earth is getting hotter because of the buildup in the atmosphere of carbon dioxide and other gases produced largely by the burning of fossil fuels. For each month this year, average global temperatures have been the highest on record.



... And almost certain to get worse

The Intergovernmental Panel on Climate Change, an international group of scientists, projects that the surface temperature of the earth could rise by about 1.8° to 6.3° F (1° to 3.5° C) by 2100. That could have serious consequences:

In the next 100 years sea levels could rise 1.6 ft. (0.5 m), threatening heavily populated coastal areas from Mississippi to Bangladesh

Extreme weather events, from hurricanes to droughts, could become more frequent and more severe

■ Warmer temperatures could foster crop production in Northern Europe and Canada but dry out important growing regions in the U.S., eastern South America and Southeast Asia

Tropical diseases like malaria may move northward and southward

💽 WHAT YOU CAN DO

Every year about 51 trillion lbs. (23 trillion kg) of carbon dioxide are released into the atmosphere. Here are examples of what you might trim from that total with a few steps:

- washers with warm or cold water. 700 lbs. (320 kg) S Forget the sport utility vehicle 2,500 lbs. per extra 10 m.p.g.

(1,134 kg per extra 4 km/l)

TIME, AUGUST 24, 1998

and buy a fuel-efficient car

The Earths Atmosphere



R Behera

Greenhouse Effect



- FACT: 15% increase in [CO₂] in last 100 years
 Cause:
 - Change from agricultural to industrial lifestyle
 - > Burning of fossil fuels (petroleum, coal)
 - > Increase CO_2 emissions (cars, factories etc...)
 - Deforestation

> Effects:

≻ Global warming





- \succ Melt polar ice caps \rightarrow flooding at sea level
- \succ Warming oceans \rightarrow more powerful storms

